Exhaled nitric oxide monitoring by quantum cascade laser: comparison with chemiluminescent and electrochemical sensors

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Abstract. Fractional exhaled nitric oxide (FENO) is considered an indicator in the diagnostics and management of asthma. In this study we present a laser-based sensor for measuring FENO. It consists of a quantum cascade laser (QCL) combined with a multi-pass cell and wavelength modulation spectroscopy for the detection of NO at the sub-part-per-billion by volume (ppbv, 1:10^{-9}) level. The characteristics and diagnostic performance of the sensor were assessed. A detection limit of 0.5 ppbv was demonstrated with a relatively simple design. The QCL-based sensor was compared with two market sensors, a chemiluminescent analyzer (NOA 280, Sievers) and a portable handheld electrochemical analyzer (MINO®, Aerocrine AB, Sweden). FENO from 20 children diagnosed with asthma and treated with inhaled corticosteroids were measured. Data were found to be clinically acceptable within 1.1 ppbv between the QCL-based sensor and chemiluminescent sensor and within 1.7 ppbv when compared to the electrochemical sensor. The QCL-based sensor was tested on healthy subjects at various expiratory flow rates for both online and offline sampling procedures. The extended NO parameters, i.e. the alveolar region, airway wall, diffusing capacity, and flux were calculated and showed a good agreement with the previously reported values. © 2012 Society of Photo-Optical Instrumentation Engineers (SPIE). [DOI: 10.1117/1.JBO.17.1.017003]

Keywords: fractional exhaled nitric oxide; breath analysis; quantum cascade laser; wavelength modulation spectroscopy; chemiluminescence; electrochemical; asthma.

1 Introduction

A method for non-invasive monitoring of inflammation from the respiratory system represents a challenge and so far, several methods have been used, i.e. exhaled breath condensate, electronic nose, and nitric oxide detection. The discovery of nitric oxide (NO) in the exhaled breath and its increased value in asthmatics paved the way for technical developments. Measurement of FENO (Fractional Exhaled NO) is known to be flow dependent. The American Thoracic Society/European Respiratory Society (ATS/ERS) statement recommends measuring FENO at a constant flow rate of 50 ml/s (FENO_{0.05}), reflecting a trade-off between sensitivity and patient comfort. However, measurement of FENO at different flow rates, so-called extended NO analysis (or known as flow-independent parameters), can be used to calculate NO parameters providing NO production from different compartments of the lung. The NO parameters are the NO from the alveolar region (C_{aw} NO) and NO flux from the airways (J_{aw} NO), derived from a linear model, and the airway wall content of NO (C_{aw} NO) and diffusing capacity of NO over the airway wall (D_{aw} NO) that can be calculated from a non-linear model. These NO parameters may give new insight into respiratory diseases.

For medical applications, there is a great requirement for a sensitive, accurate, compact, convenient and inexpensive sensor. Several instruments are now available making use of either chemiluminescence, electrochemical, or laser-based technologies. Care has to be given to validate and calibrate each device to avoid conflicting reports between studies. Chemiluminescent detection of FENO is considered by many researchers as the "gold standard." It provides accuracy and precision but is bulky and expensive, high on-going running costs, and also requires technical expertise in calibration, limiting their use in routine patient care. Electrochemical sensors are convenient for development of portable hand-held analysers. However, the reproducibility of FENO measurements for absolute values is still subject to conflicting reports and can have consequences of relying on a single estimate of the level of FENO in driving clinical management. The advances of quantum cascade laser (QCL) technologies have opened up new opportunities for novel mid-infrared (mid-IR) gas sensors. The QCL-based sensors are well suited for mid-infrared spectroscopic trace gas sensing due to their narrow linewidth, high power at room temperature, and continuous wave (CW) operation at mid-IR wavelengths (3 to 24 μm). Several approaches for the optical sensing of NO have been reported. Sensors based on absorption spectroscopy using a multi-pass cell or a high finesse cavity for cavity enhanced or cavity ring-down spectroscopy, photoacoustic spectroscopy, and Faraday modulation spectroscopy have been successfully
implemented to reach a detection sensitivity in the order of single part per billion by volume (ppbv, $1 \times 10^{-9}$) and below.

In this paper, we report on the development and performance evaluation of a laser-based NO sensor, utilizing a CW, thermoelectrically cooled QCL as a light source. The sensor was validated for $F_{E}$NO measurements: linearity, selectivity, precision, accuracy, and detection limit were determined. The QCL-based sensor was tested on healthy adults and asthmatic children. Performance comparisons were done with two other techniques, namely chemiluminescence and electrochemical sensor.

2 Experimental

2.1 Gas Sensors

The QCL-based NO sensor uses wavelength modulation spectroscopy technique (WMS) and consists of a continuous-wave distributed feedback quantum cascade laser (CW-DFB QCL) operating in the wavelength region of 5.2 μm (1891 to 1908 cm$^{-1}$), a multi-pass cell, and a room temperature detector. The laser beam is sent through an astigmatic Herriott absorption cell of 400 ml (AMAC-76, Aerodyne Research, USA), offering a total optical path length of 76 m and which consists of two concave mirrors where light is reflected multiple times, enhancing the effective path length through the breath sample. The absorbed amount of light at the output of the cell is proportional to the NO concentration present in the cell. The detection is performed by a photovoltaic detector (PV-6, Vigo Systems) working at room-temperature, thus eliminating the need for a highly sensitive liquid nitrogen-cooled IR detector, simplifying daily use of the system and allowing long-term automated operation. In order to improve sensitivity toward the NO target limit of ≤1 ppbv required for breath analysis, WMS is implemented by modulating the injection current of the laser at a frequency of 100 kHz. The absorption signal is transposed into a frequency domain where noise sources are weaker, and absorption spectra is obtained by demodulating the detected signal at the second harmonic ($2f$) using a lock-in amplifier (model SR844, Stanford Research Systems) with the time constant set to 100 μs. This approach allows a sensitivity of 0.5 ppbv within 1 s averaging. To reach a fast response of the system, a pressure of 70 cmH$_2$O is maintained in the multi-pass cell and a sample flow rate of 900 ml/min is generated. Therefore, the cell is refreshed in less than 2 s, making the sensor suitable for online monitoring of exhaled breath. A schematic arrangement of the experimental setup is shown in Fig. 1. The QCL-based sensor was calibrated with a reference mixture of 100 ± 3 ppmv of NO in N$_2$ (prepared by VSL-National Dutch Metrology Institute) before analysis of a breath sample. A N$_2$ gas bottle was used as a NO-free gas reference.

In this study, the performances of the QCL-sensor were investigated. To do that, 11 gas mixtures were prepared from our reference gas mixture (100 ppbv NO in N$_2$) diluted in pure N$_2$ to cover the range 10 to 100 ppbv, and 14 gas mixtures were made from a 20 ppm ± 2% bottle of NO for concentrations up to 4 ppm. The different concentrations of NO were produced by using two mass flow controllers (Brooks Instrument, max flow: 25 l/h and 5 l/h with an accuracy of ±1%).

Contrary to the QCL sensor which directly measures the NO concentration, the chemiluminescence system is based on the reaction between NO and O$_3$, which generates NO$_2$ in the exited form. When NO$_2$ returns to a stable state, light is emitted. The amount of light, which is proportional to the amount of NO, is measured by a photomultiplier. Calibration of the Sievers NOA 280 was performed each day prior to use with the same mixture of NO as with the QCL-based sensor, i.e. with N$_2$ gas bottle and 100 ± 3 ppbv of NO in N$_2$. The sample flow into the Sievers is 200 ml/min. The chemiluminescence analyzer is connected to a computer with the NO analysis software, which provides a graphical display of $F_{E}$NO concentration during the analysis process.

Despite the chemiluminescence NO sensor being the gold standard reference for NO measurements in breath, its use for routine clinical practice is limited by its size and expense. Portable hand-held, relatively inexpensive NO analysers based on electrochemical analysis have been introduced on the market a few years ago. The NIOX MINO (Aerocrine AB, Sweden) Asthma Inflammation Monitor is one of the available sensors. This sensor is ideally suited for use in primary care, where the majority of asthma patients are managed. In this study, the accuracy of NIOX MINO measurements was assessed. The manufacturer stated an accuracy of ±5 ppbv of measured value below 50 ppbv and ±10% at or above 50 ppbv.

2.2 BreathCollection

The QCL-based sensor was used for $F_{E}$NO monitoring with both online and offline breath-collection.

To perform online measurements, a commercially available breath sampler (Loccioni, Italy) was used to monitor the pressure of the exhaled breath within an acceptable range and to measure the breath CO$_2$ concentration level. The breath sampler meets the American Thoracic Guidelines (ATS) for collecting breath by providing a back pressure of 10 cmH$_2$O, to ensure soft palate closure and prevent nasal contamination, and allowing the patient to maintain constant exhalation flow.
The exhalation flow ranged from 15 to 250 ml/s. The target flow rate was maintained within 5%.

During the sampling, the collected breath entered the Loccioni breath sampler. The CO2 concentration profile and airway pressure were simultaneously displayed in graphical forms on the sampler display. When the CO2 concentration reached 3%, a part of the breath was sent to the NO sensor. The 3% value has been chosen since it is the start of the CO2 plateau for most of people. The FENO level in ppbv was acquired and plotted in real-time on a computer screen.

In addition, the breath sampling pipe was heated to ≥38°C to prevent water condensation. A typical recorded signal at flow 50 ml/s is shown in Fig. 2. FENO level was sampled after the estimated start of the CO2 plateau region. The end-tidal of the CO2 exhalation trend determines the point where the NO plateau is measured, i.e. 5 s before the end-tidal CO2 point. One breath sample was collected at each flow rate (15, 50, 100 and 250 ml/s) from each patient.

Offline collections were also performed as samples can be collected at a distant collection site from the NO sensor. A custom-built breath-collection device was used to collect single breaths into bags with the subject exhaling at specific constant flow rates. The breath-collection device was based on the guidelines of the ATS for the sampling of exhaled NO; it is very simple, inexpensive, and has been tested in previous studies. The exhaled breath line consists of a mouth-piece connected to a discard bag (400 ml) and an NO-impermeable aluminum-foil air bag of 500 ml capacity (Mylar balloon, ABC ballonnen, Zeist, The Netherlands). To maintain a constant exhalation flow, the mouth pressure was monitored by the patient during the sampling process. Breath was collected at various constant flow rates from each subject by changing the resistance of the breath line to maintain a mouth pressure of 10 cmH2O.

To assess the agreement of the two sampling procedures, offline FENO measurements in comparison to online measurements, 23 healthy individuals performed online and offline single breath maneuvers at a flow of 50 ml/s. To prevent systematic errors, 11 persons did first online followed by offline breath maneuvers, and 12 in the reverse order.

2.3 Extended NO Analysis

Different exhalation flow rates can be performed with both online and offline exhaled breath measurements by changing the resistance of the sampling line and keeping a back pressure of 10 cmH2O during the exhalation. Measuring the NO plateau for multiple flow rates allows an estimation of the NO extended parameters, which may provide additional useful clinical information. The simple two-compartment model describes exhaled NO arising from two compartments: the airways and the alveolar region. It is based on three flow-independent exchange parameters, one describing the steady-state NO alveolar concentration (CwNO, ppbv), and two describing the airway region (the airway NO diffusing capacity (DawNO, pl·s−1·ppbv−1) and the airway wall NO concentration (CawNO, ppbv)). The potential of these parameters lies in their ability to split exhaled NO into two important anatomic subdivisions of the lungs and also to provide both structural and metabolic information relevant to the NO pathways. To determine the airway and alveolar contribution to exhaled NO, multiple exhalation flow rates must be accomplished. Different approaches are described in the literature and recently used in combination with a QCL-based sensor to determine extended NO parameters in chronic obstructive pulmonary disease. We selected the method described by Höggan et al. as it requires only three exhalation flows. This method is based on a non-linear model with a quality control to notify erroneous NO values. NO values from different flow rates are included into a software program with a second order algorithm, which will render values of CwNO, CawNO and DNO.

We measured FENO at different flow rates controlled by the Loccioni sampler: a low exhalation flow (15 ml/s), a medium flow (100 ml/s), and a high flow rate (250 ml/s). To validate the sampling system, the exhalation flow of 50 ml/s is also measured (FENO0.05) and compared to calculated value by the model. A group of 20 subjects performed online single breath maneuvers at four expiratory flows as mentioned previously. After each maneuver, the subject was given a 1-minute pause before performing the next exhalation at a different flow.

3 Results

3.1 QCL-Based Sensor Performance

Prior to the study on patients, the performance of our QCL-sensor was evaluated. With the prepared NO concentrations, the QCL-sensor was tested in terms of linearity, precision, accuracy, sensitivity, and selectivity. To minimize random error, the

<table>
<thead>
<tr>
<th>Table 1</th>
<th>Main characteristics of the QCL-sensor.</th>
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<tbody>
<tr>
<td>Sensitivity</td>
<td>&lt;1 ppbv in 1 s</td>
</tr>
<tr>
<td>Repeatability (Precision)</td>
<td>±5%</td>
</tr>
<tr>
<td>Accuracy</td>
<td>99%–104%</td>
</tr>
<tr>
<td>Dynamic Range</td>
<td>1 ppbv–4 ppmv</td>
</tr>
<tr>
<td>Response Time</td>
<td>Log Time: 2 s</td>
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</tbody>
</table>
concentrations were measured six times. One measurement consists of flushing the absorption cell for 5 min with the gas and by taking the last value displayed by our sensor. Table 1 summarizes the main characteristics of the sensor. The linearity response of the system, Fig. 3, is given by the coefficient of determination ($R^2 = 0.998$). Precision of the sensor was evaluated in conditions of their repeatability, and accuracy was determined by comparing the mixtures and the measured concentration. The calculated accuracy was between 99 and 104% and is limited by the accuracy of the mass flow controllers used for the dilution. The precision of the system expressed as RSD (relative standard deviation) was below 5%.

The best minimum detectable NO concentration achieved with QCL-based sensor with a 1 s averaging time is 0.5 ppbv. A better sensitivity can be reached by increasing the integration time. An Allan variance analysis of the data, which presents the variance of the data as a function of integration time of the sampling, is shown in Fig. 4. It displays the reduction in noise level as the integration time is increased.

As exhaled breath carries a number of compounds at high concentration such as water and CO$_2$, the selectivity of the sensor is important. The emission spectrum of the QCL allows accessing of the NO absorption line within the fundamental vibration band without interference of water and CO$_2$, as shown in the HITRAN simulation. The response of the QCL sensor to humid air and 5% CO$_2$ has been experimentally investigated and has provided good agreement with the HITRAN calculations.

### 3.2 Offline/Online Comparison

The QCL-based sensor can perform F$_{E}$NO measurements online or offline. If the advantages of online analysis are obvious, the benefits of offline analysis are apparent when analyzing large sample numbers, for example during clinical studies. To examine the diagnostic capability of our sampling procedure, online and offline measurements in the same subjects are compared, using the Bland-Altman method. The Bland-Altman plot, Fig. 5, consists of the difference between a pair of measurements (in this case offline–online) versus their average. On average, the difference in NO concentration between the two sampling methods (offline-online) was $-0.1$ ppbv and the limit of agreements was $\pm 1.0$ ppbv. By measuring the NO concentration in bags within 24 h of collection, no significant differences or combined effects of measurement technique or flow rate were noted. The sample is reportedly stable for up to 24 h, but it was observed that NO concentrations in bags decreased over time up to 7% after 48 h (data not shown).

### 3.3 Validation of QCL-Based Sensor for Multiple Flows Analysis

For each individual, the three independent parameters are calculated. Figure 6 shows the data as medians with lower and upper quartiles (25 to 75%). The level of agreement between the measured and calculated $F_{E}$NO$_{0.05}$ is determined by the coefficient of determination $R^2$ (average between calculated and measured: $-0.4$ ppbv, $SD = 0.72$, $R^2 = 0.9997$). Data
are also compared to previous results, and these results are in agreement with values presented by Högman et al. Regardless of the lack of agreement about what flow rates to use to calculate extended NO parameters, those results demonstrate the validity of our sampling procedure and confirm the efficiency of the theoretical model to give values which are consistent.

3.4 Comparison of the QCL-Based Sensor with Chemiluminescence and Electrochemical Sensors

\( F_2\text{NO} \) from 22 asthmatic children (age range: 6 to 16 years) was measured with three sensors. They provided offline samples in aluminum bags for the QCL-based sensor and the chemiluminescence, respectively, at flow 50 ml/s, followed by online \( F_2\text{NO} \) sample measured by the NIOX MINO two minutes later. Ambient NO varied between 1 and 4 ppbv over the study. The bags were measured simultaneously with the QCL-based and the chemiluminescent sensors on the same day as the sampling. Measurements were achieved with a 1 s integration time for both devices. The Bland-Altman plot, Fig. 7(a), shows a high degree of agreement between the QCL-based sensor and the chemiluminescence device with an average difference of \(-0.1\) ppbv and a standard deviation of 1.1 ppbv.

The comparison between the QCL-based sensor and the NIOX MINO, Fig. 7(b), shows an average value of 1.6 ppbv with a SD of 1.7 ppbv.

4 Discussion

This study reports a QCL sensor, combining a multi-pass cell and wavelength modulation spectroscopy, suitable for \( F_2\text{NO} \) measurements. The achieved sensitivity of 0.5 ppbv within 1 s integration time, appropriate for \( F_2\text{NO} \) measurements, is comparable with other on-going optical sensor developments and does not represent the ultimate sensitivity. The fast response time achieved (<2 s), despite the use of a multi-pass cell, allows online analysis by providing enough time resolution to follow NO variations during a single breath, as well as offline measurements. Although presently used in a setup configuration, the QCL sensor has a real potential to be developed and integrated into a compact and convenient device configured for autonomous operation in clinical applications.

For the present work, the obtained results are directly comparable with those obtained by the chemiluminescent NIOX sensor, the device currently considered as the gold standard for measuring \( F_2\text{NO} \). A Bland-Altman plot demonstrated good agreement between both devices for a wide range of NO concentrations, making the system trustable on any asthmatic patient regardless of the degree of airway inflammation. When coupled with a commercial or home-made sampler, single or multiple flow rates can be easily performed. The only requirement is to be able to monitor CO\(_2\), the flow rate, and mouth pressure during the exhalation. The system offers a wide number of configurations fitting to each study, including extended NO analysis. The possibility of measuring multiple flow rates was demonstrated in healthy subjects by successfully comparing calculated extended NO parameters (from \( F_2\text{NO}_{0.015} \), \( F_2\text{NO}_{0.10} \), \( F_2\text{NO}_{0.25} \)) with previous papers. In addition, the calculated \( F_2\text{NO}_{0.05} \) from the model is in agreement with the measured \( F_2\text{NO}_{0.05} \). By adapting the sampler device, any flow...
rate can be used with the QCL sensor. A recent study also demonstrated the efficiency of a QCL-based sensor for multiple flow rates analysis in diseases other than asthma, as in chronic obstructive pulmonary disease.37

We also found a statistically significant difference (p-value = 0.0003) between FE NO values measured by the QCL-sensor and the NIOX MINO. Ekroos et al.42 reported on the exhaled NO from a selected group of healthy non-smoker adults and assessed their long-term variation. The variation of NO concentrations (95% confidence interval) was 1.1 ± 1 ppb within the interval of seven days. This is in agreement with the difference in the methods we report here. When patients start treatment this effect is much larger. Silkoff et al.43 showed that the reduction in exhaled NO was around 40%. Therefore, we conclude that the difference between the two methods is not medically relevant for the measurement of F ENO0.05.

The validity of the NIOX MINO has been subject of conflicting papers. Some studies reported strong correlations and excellent reproducibility of the absolute value of FE NO measurements obtained using the NIOX MINO.34,48 They support the recommendation of performing only one measurement. Other studies did not support that conclusion.36,46 Our data suggests that even by performing one FE NO measurement, the level of agreement between the gold standard device, the QCL-sensor, and the NIOX MINO is certainly clinically acceptable. However, the 5 ppbv sensitivity prevents its use in determining the extended NO parameters.

In conclusion, this study shows that there is clinically acceptable agreement between the three main technologies used for the development of NO sensors. Whereas chemiluminescence analysers tend to be expensive, large and poorly portable, laser-based systems and electrochemical sensors demonstrate interesting opportunities to make FE NO measurements in the primary care. Compared to the NIOX MINO and generally to electrochemical sensors, QCL-based systems, using highly selective optical spectroscopy principles, offer higher sensitivities and have already proven their advantages as a trustable and accurate technique over time. They are also suitable for multiple flow rates analysis as they can perform analysis at any flow without modification of the sensor. Compact devices based on laser spectroscopy are presently used in many fields of life sciences, and it is a matter of time before they will be used in any hospital for diagnosis, monitoring, and control of diseases such as asthma at single or multiple flow rates.

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